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Emissions from NH₃ Combustion Engines

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Contents

- Introduction to emissions from NH₃ combustion.
- Mechanisms of emissions formation from NH₃ combustion.
- Emissions abatement and control for NH₃ exhaust.
- Future directions of NH₃ exhaust emissions control.

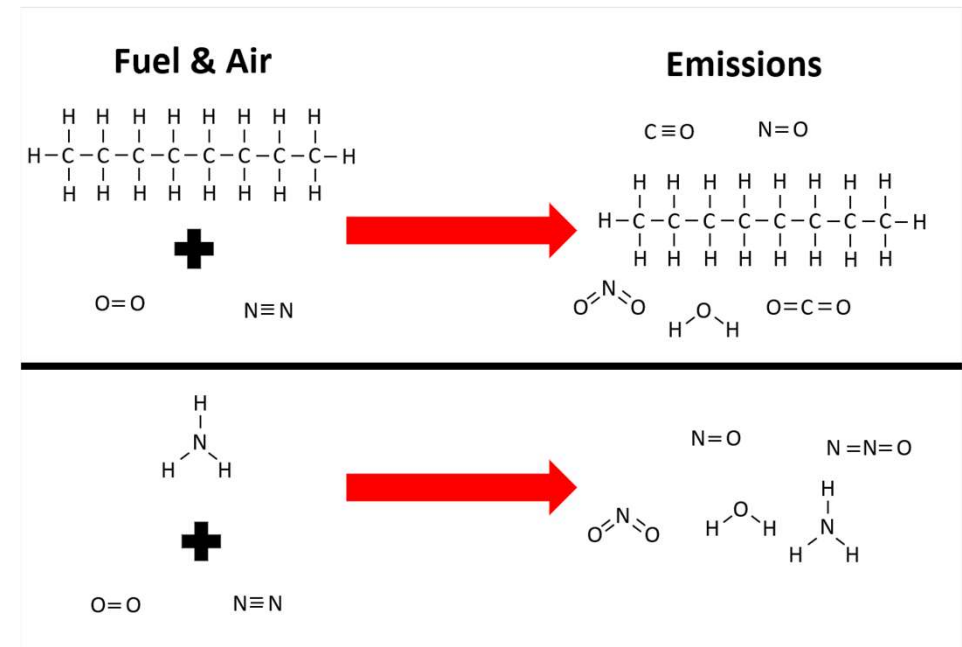
Emissions from NH_3 Combustion

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How do emissions from NH₃ differ to conventional fossil-based fuels

- NH₃ is a carbon-free fuel meaning there are no CO or CO₂ emissions produced.
- As NH₃ is not a HC there will be no HC emissions (other than any emitted from the lubrication oil).
- Unburnt NH₃ that did not partake in the combustion process (slip ammonia).
- NO_x emissions are still prevalent from NH₃ combustion and present a challenge.



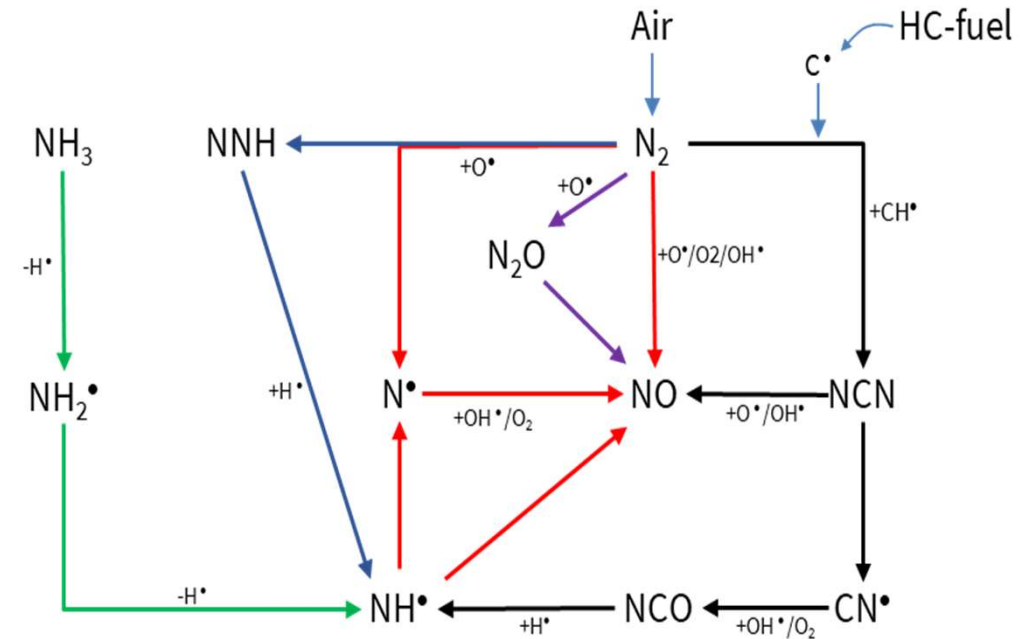
Emissions produced from combustion of gasoline (iso-octane) and ammonia.

Formation of NO_x through NH_3 combustion

- NO_x production is done through 5 main pathways:

- Thermal (T-sensitive)
- Fuel (N in fuel sensitive)
- Prompt (fuel rich, $T < 1800 \text{ K}$)
- N_2O intermediate (fuel lean, $T < 1800 \text{ K}$, $\uparrow P$)
- NNH mechanism (H sensitive)

- Due to the nitrogen present within NH_3 the fuel NO_x pathway becomes important.
- The lower adiabatic flame temperature of NH_3 compared to gasoline reduces the contribution from thermal NO_x .



Simplified scheme of formation pathways of the NO species by different mechanisms

Emissions Abatement for NH_3 Combustion

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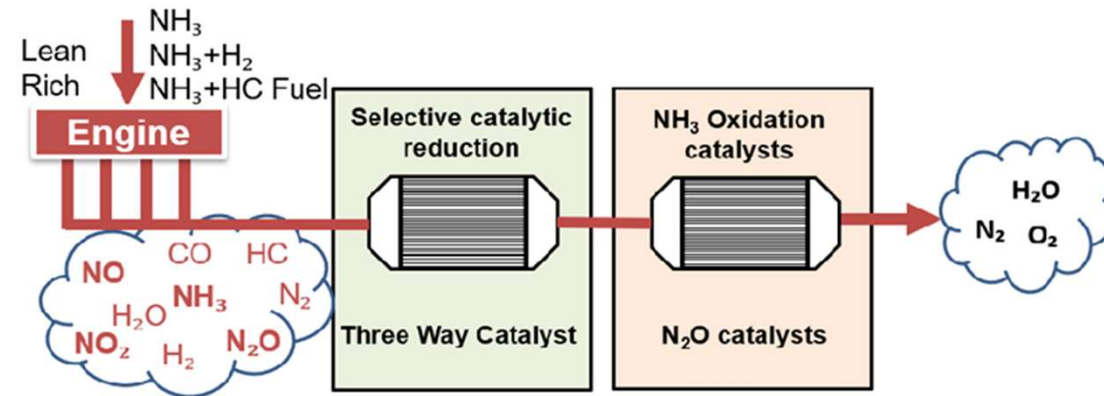


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Aftertreatment for NH₃ exhaust

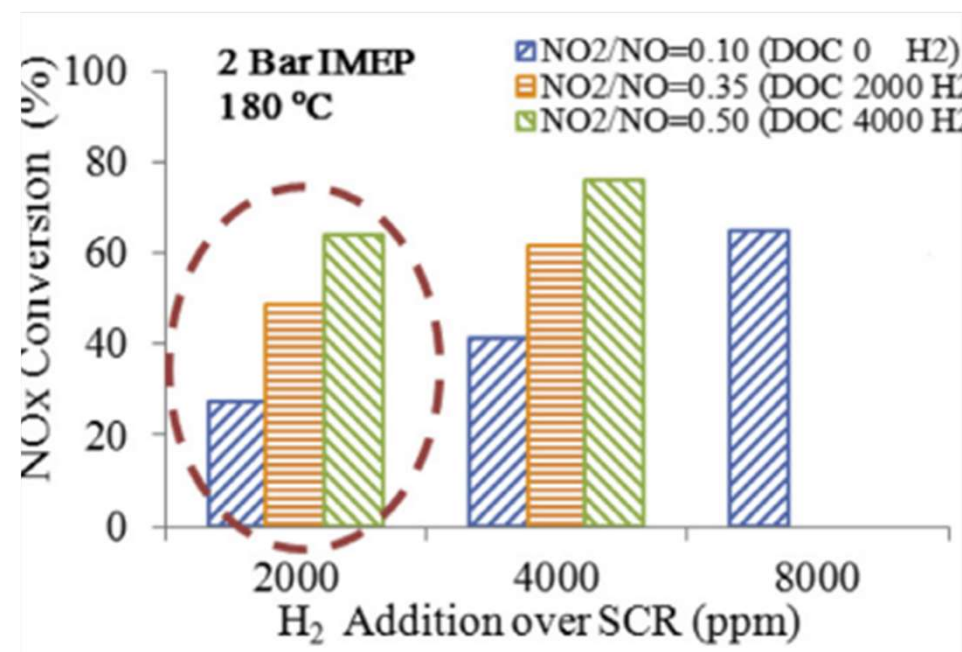
- The different composition of NH₃ exhaust gas produces different challenges to aftertreatment systems than conventional gasoline exhaust gas.
- There are several options for catalytic aftertreatment systems applicable to NH₃ exhaust gas control:
 - Selective catalytic reduction (SCR)
 - Ammonia slip catalyst (ASC)
 - N₂O catalyst



Aftertreatment solutions for ammonia combustion engines.

Selective catalytic reduction (SCR)

- Removes NO and NO₂ through reduction reactions with a suitable reductant.
- Reductants include:
 - Ammonia
 - Hydrogen
 - Hydrocarbons (alcohols, alkanes, etc.)
- If there is not a sufficient number of reductants available in the exhaust gas they must be added through injection.
- If there are too many reductants, they will slip through the SCR.



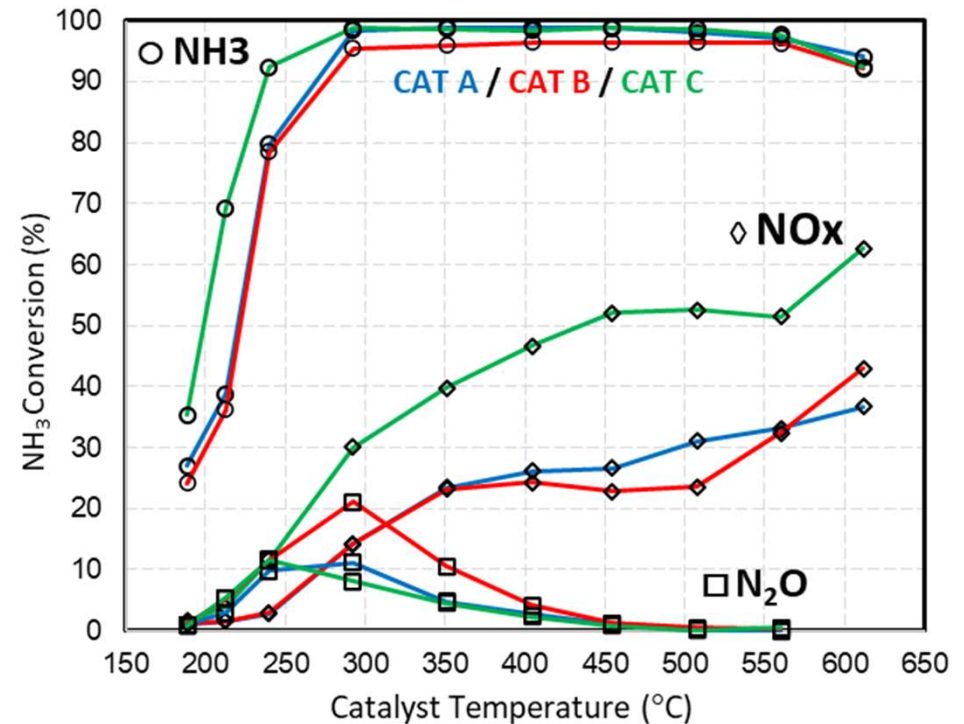
Effect of H₂ addition on SCR NO_x conversion efficiency. (Chemical Engineering Journal 270 582–589.)

Ammonia slip catalyst (ASC)

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- Excess NH₃ from combustion or SCR dosage is an issue.
- NH₃ slip catalysts have been studied when paired with SCR catalysts to address over dosage.
- There is secondary production of N₂O and NO_x as NH₃ is oxidized.
 - $2\text{NH}_3 + 1.5\text{O}_2 \rightarrow \text{N}_2 + 3\text{H}_2\text{O}$
 - $2\text{NH}_3 + 2\text{O}_2 \rightarrow \text{N}_2\text{O} + 3\text{H}_2\text{O}$
 - $2\text{NH}_3 + 2.5\text{O}_2 \rightarrow 2\text{NO} + 3\text{H}_2\text{O}$



NH₃ conversion & products. (SAE Transactions, Vol. 116, Section 4: JOURNAL OF FUELS AND LUBRICANTS 182-186)

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N₂O catalyst

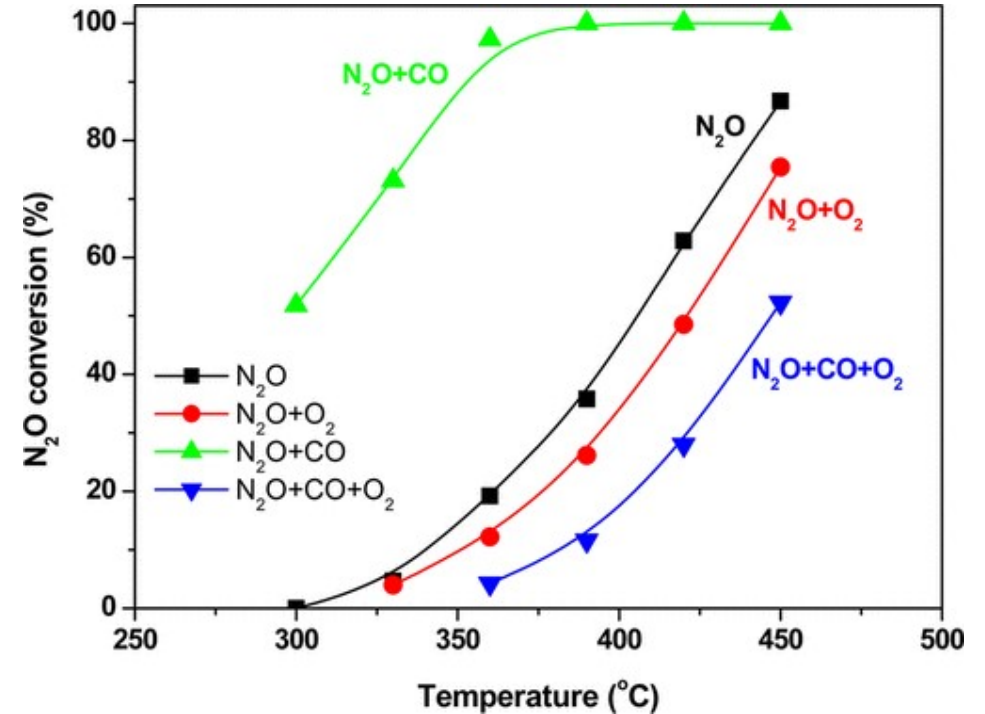
- N₂O decomposition catalysts break down N₂O into N₂ and O₂.



- $N_2O + O^* \rightarrow N_2 + O_2$
- Nobel metal based: High cost with sensitivity to H₂O/O₂ sensitivity.

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N₂O conversion – (Applied Catalysis B: Environmental 75, 167–174)

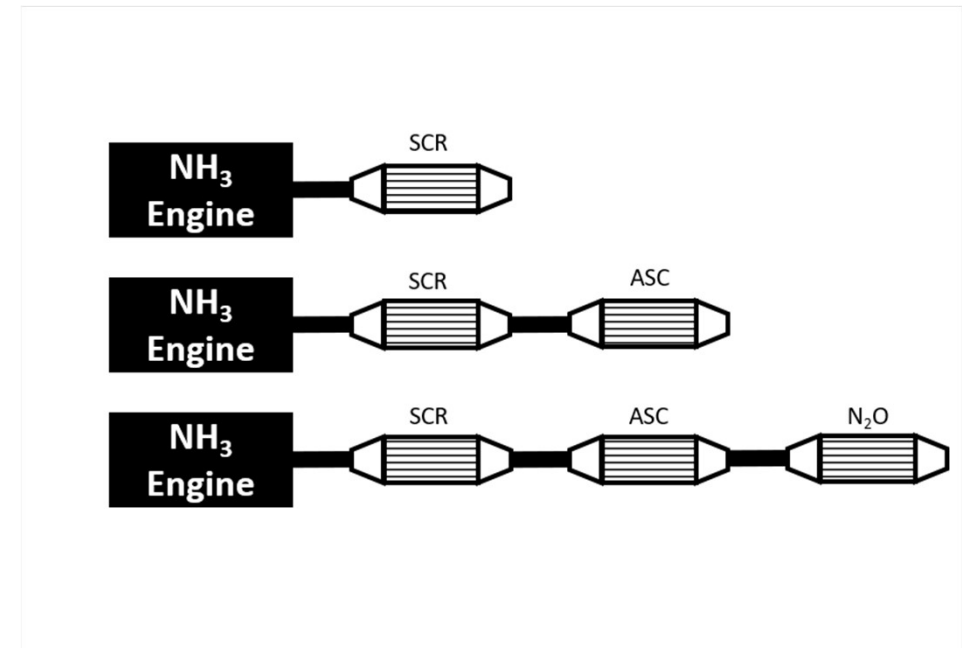
Summary & Future Directions

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Summary and future directions

- Aftertreatment for NH₃ combustion requires different catalytic solutions compared to conventional fossil-fueled combustion.
- Main challenges are NH₃ slip and emissions of NO_x and N₂O.
- Catalytic aftertreatment systems can provide a solution to this challenge but further research is required.
- There will be a trade off between emissions, cost and practicality.



Different proposed aftertreatment architecture for NH₃ engines.

Many Thanks

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