

# MariNH<sub>3</sub>

Clean, green ammonia  
engines for maritime

## Towards a Sustainable Decarbonised Future: Emissions formation and abatement from Ammonia Fuelled Engines

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# Contents

- Introduction to emissions from NH<sub>3</sub> combustion.
- N-emissions formation.
- After-treatment systems and control for NH<sub>3</sub> exhaust.

# Introduction

- $\text{NH}_3$ : Characterized by a **strong Irritant odor** and is **highly toxic** to the human body.
- $\text{NO}_x$ : Contributes to **photochemical smog, acid rain, and air pollution**; it is also **highly toxic**.
- $\text{N}_2\text{O}$ : A **long-lived greenhouse gas (GHG)** and an **ozone-depleting substance**.



Zero carbon Fuels ( $\text{H}_2/\text{NH}_3$ )

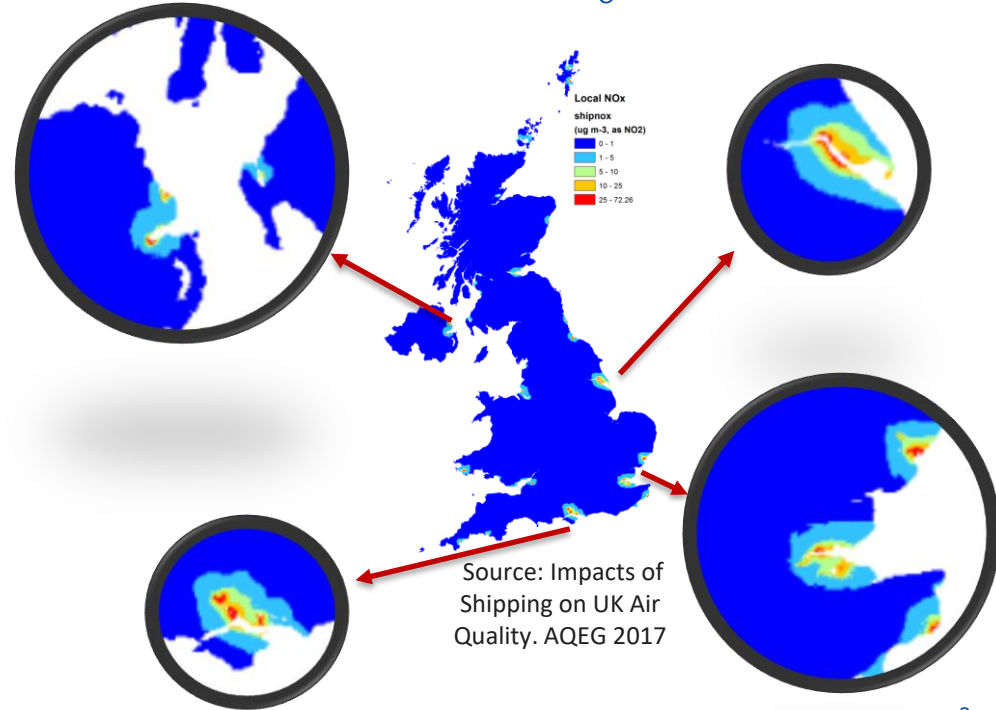
$\text{CO} + \text{THC} + \text{NO}_x$

$\text{NH}_3 + \text{NO} + \text{NO}_2 + \text{N}_2\text{O}$

Conventional Fuel

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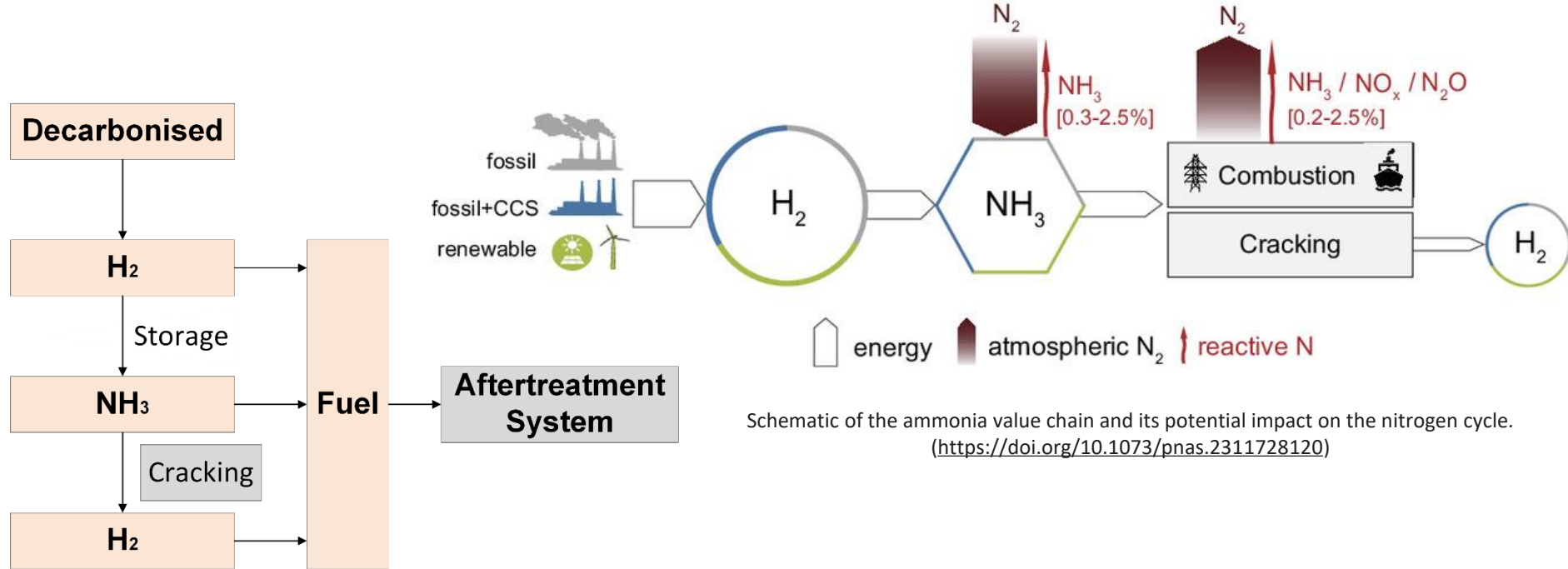
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# Introduction

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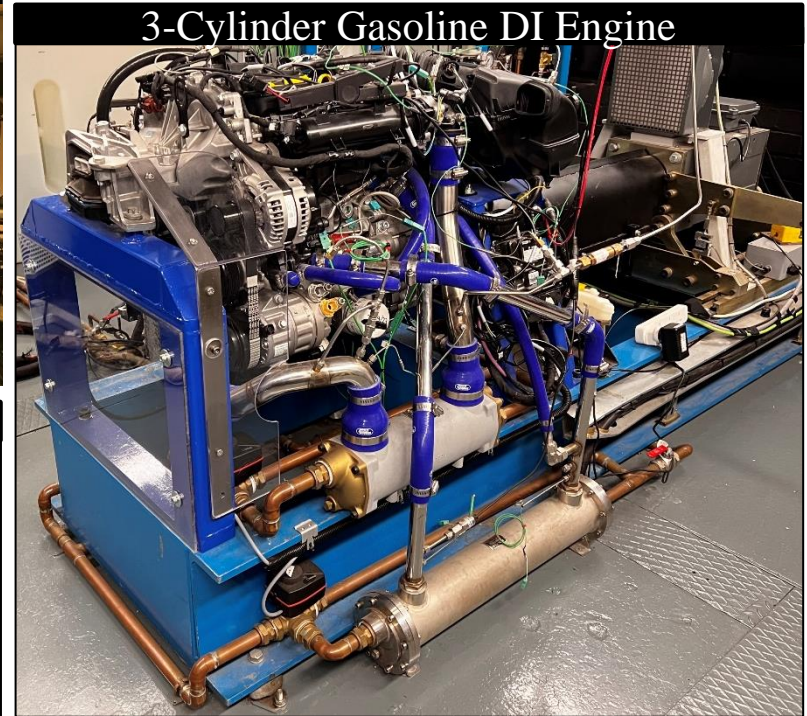
Schematic of the ammonia value chain and its potential impact on the nitrogen cycle.  
(<https://doi.org/10.1073/pnas.2311728120>)



# Experimental facilities

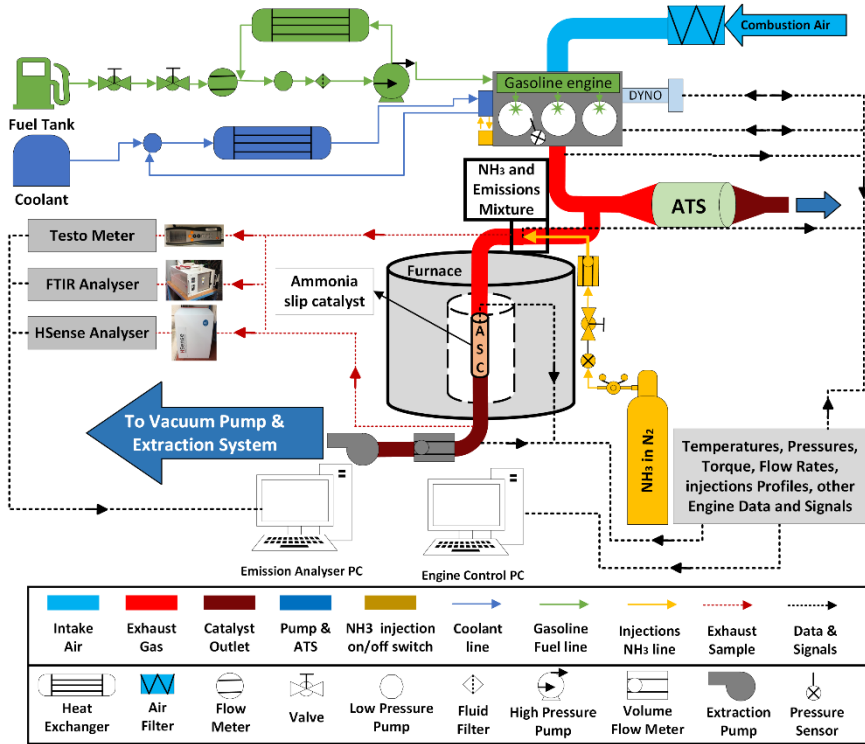
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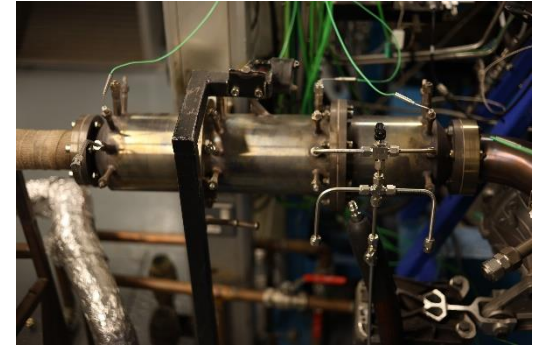
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# Experimental facilities



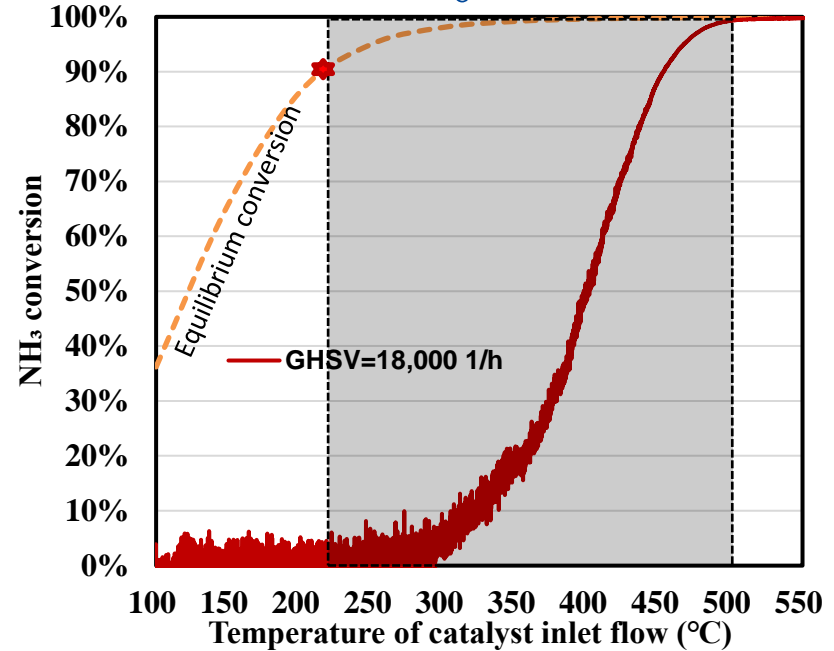
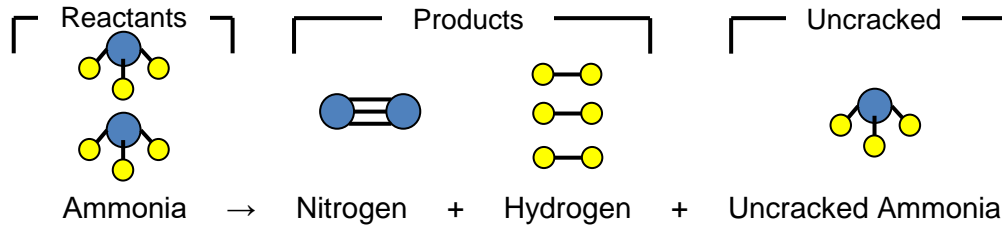
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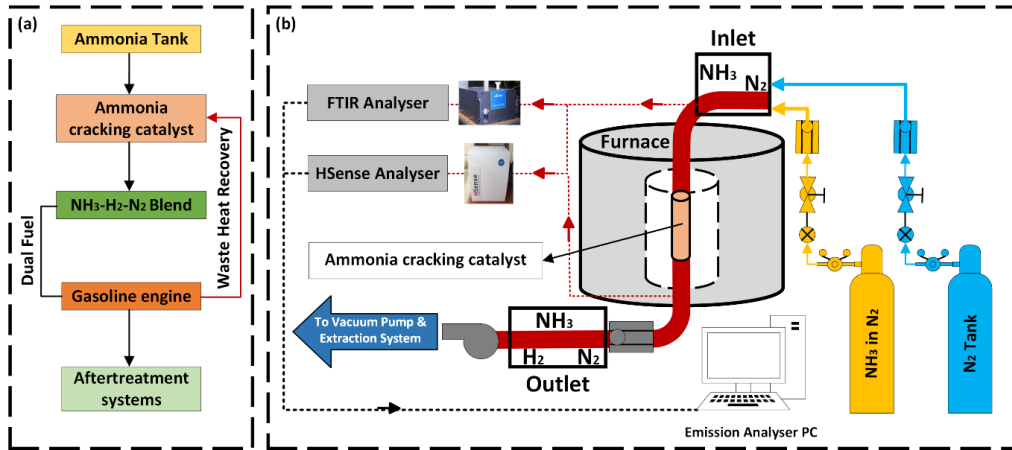
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# Ammonia cracking



Ammonia Conversion Laboratory Results for the University of Birmingham (Accepted for publication in Thiesel 2024)

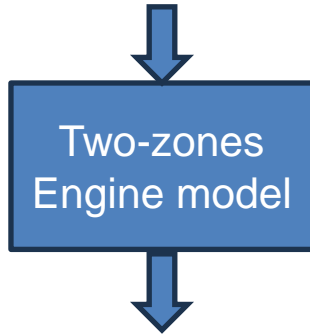
Ammonia cracking catalyst system scheme (Accepted for publication in Thiesel 2024)



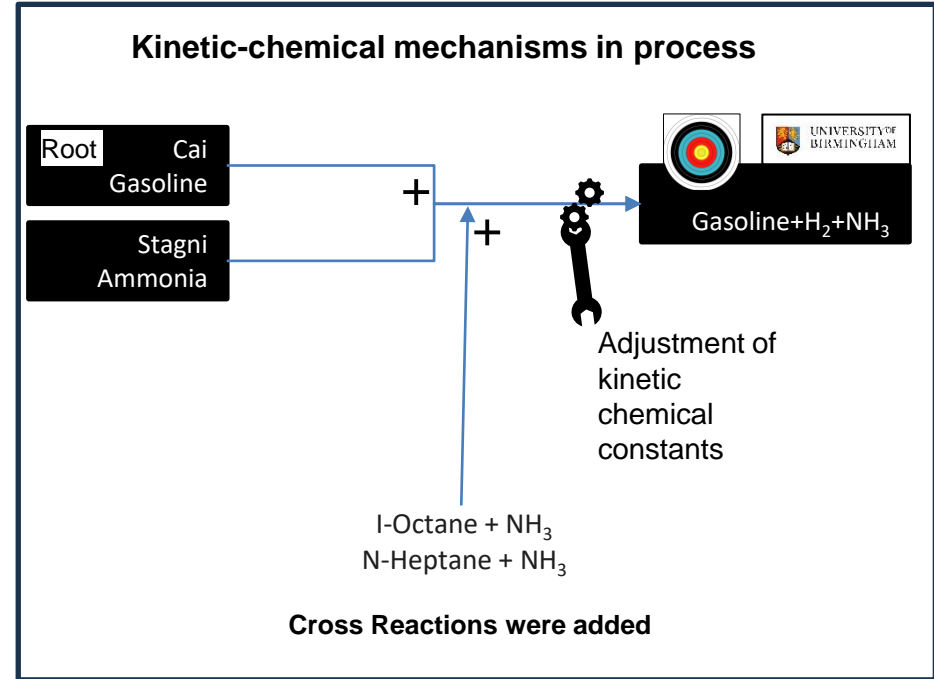
# N-emissions formation

## Fundamentals modelling

- Conditions ( $P_0$   $T_0$ ,  $\lambda$ ),
- P vs. CAD data,
- Engine geometric data



- Emissions ( $\text{NO}$ ,  $\text{N}_2\text{O}$ ,  $\text{NO}_2$ )
- Slip ( $\text{H}_2$ ,  $\text{NH}_3$ )
- Engine Performance (Torque)

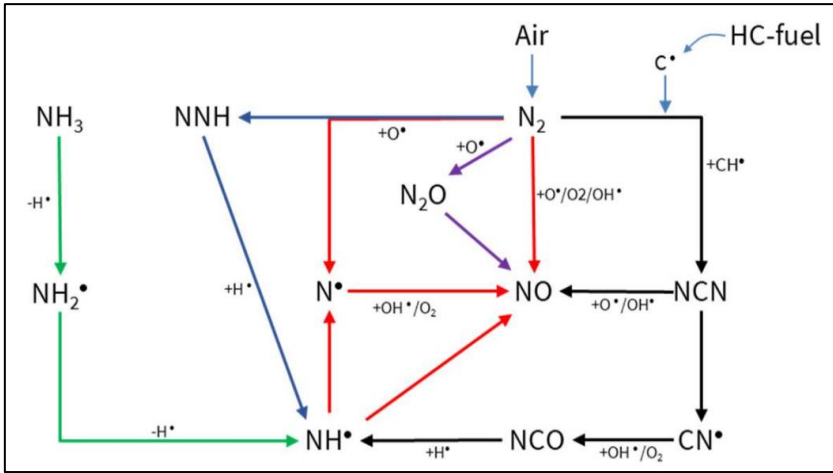




# N-emissions formation

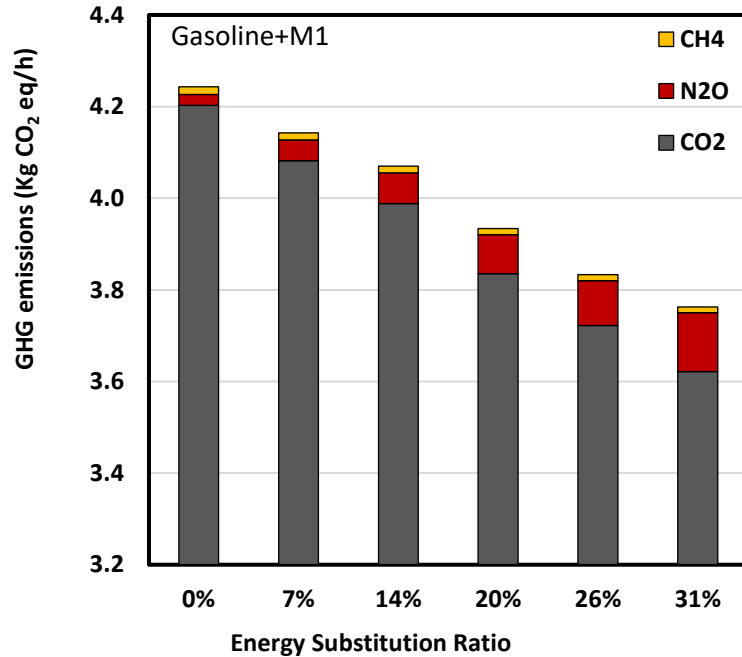
- NO<sub>x</sub> production is done through 5 main pathways:
  - ➔ Thermal (T-sensitive)
  - ➔ Fuel (N in fuel sensitive)
  - ➔ Prompt (fuel rich, T < 1800 K)
  - ➔ N<sub>2</sub>O intermediate (fuel lean, T < 1800 K, ↑ P)
  - ➔ NNH mechanism (H sensitive)

- Due to the nitrogen present within NH<sub>3</sub> the fuel NO<sub>x</sub> pathway becomes important.
- The lower adiabatic flame temperature of NH<sub>3</sub> compared to gasoline reduces the contribution from thermal NO<sub>x</sub>.



Simplified scheme of formation pathways of the NO species by different mechanisms

# N-emissions formation

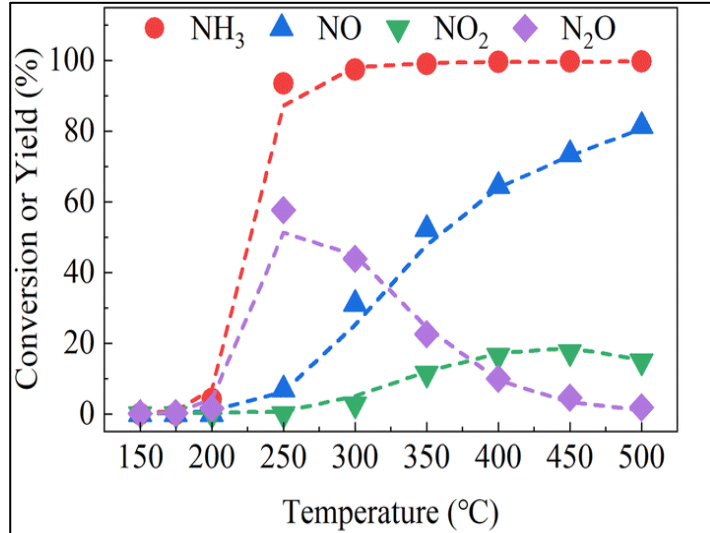


- CO<sub>2</sub> emissions continue to decrease.
- The potential contribution of the greenhouse gas **N<sub>2</sub>O increases.**
- There was a consistent decrease in the total potential greenhouse gas emissions.

Green House Emissions as the ESR increases from a SI engine fuelled with Gasoline+M1 (5%NH<sub>3</sub>/70%H<sub>2</sub>/25%N<sub>2</sub>) (Accepted for publication in Thiesel 2024)

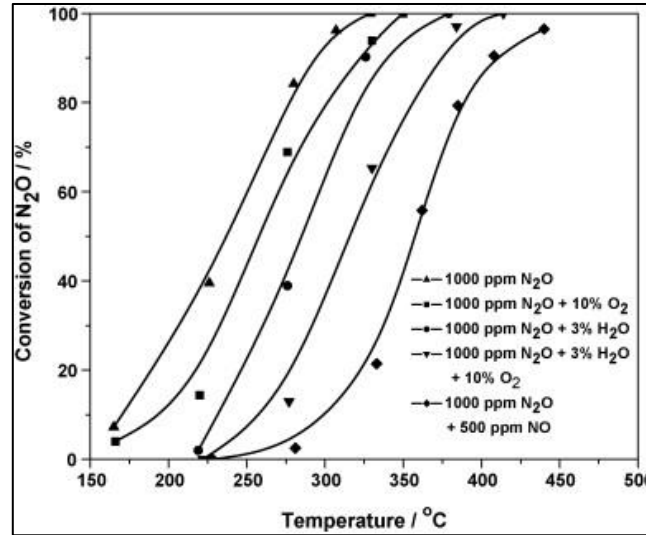
# After-treatment

## Ammonia slip catalysts (ASC)

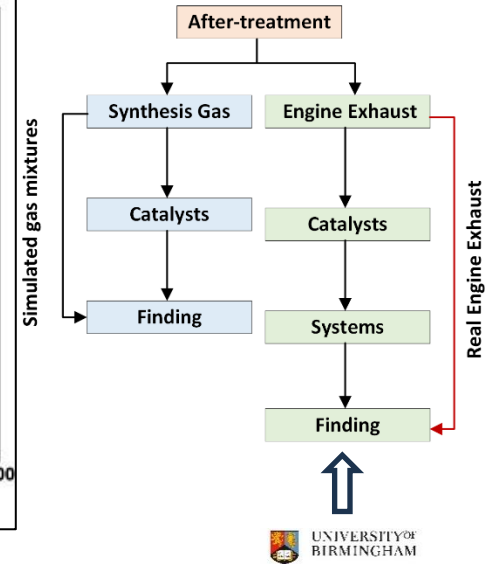


NH<sub>3</sub> conversion and product yields under NH<sub>3</sub> oxidation conditions. Yao D, Li Y, Wu F, et al. Reaction Chemistry & Engineering, 2023, 8(8): 2040-2051.

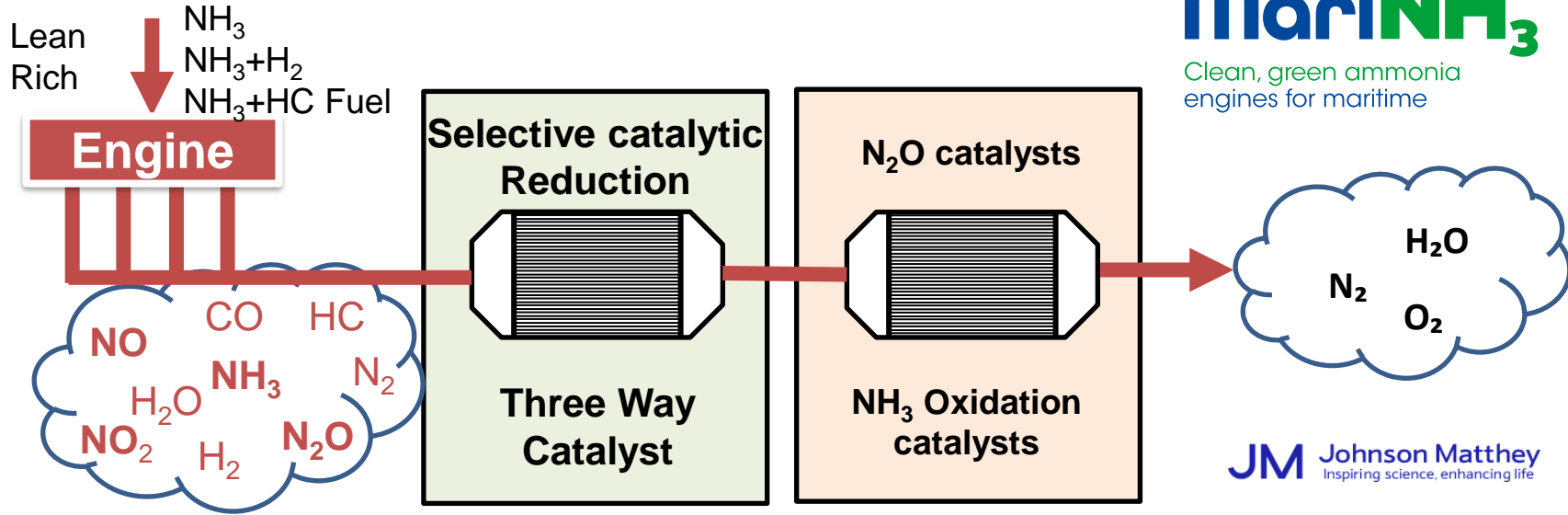
## De-N<sub>2</sub>O catalyst



N<sub>2</sub>O conversion – (Applied Catalysis B: Environmental 75, 167–174.)



# After-treatment



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- $\text{NH}_3$  as a fuel in ICE leads to complex unburnt  $\text{NH}_3$  -  $\text{NO}_x$  -  $\text{N}_2\text{O}$  trade-offs.
- The different composition of  $\text{NH}_3$  exhaust gas produces different challenges to aftertreatment systems than conventional gasoline exhaust gas.
- Exhaust after-treatment required to help with combustion trade-offs leading to combination of catalytic components.

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# Many Thanks

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